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1,3-Bis(2-pyridylaminomethyl)benzene

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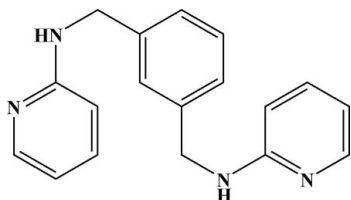
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Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.044; wR factor = 0.116; data-to-parameter ratio = 17.5.

The complete molecule of the title compound, $\text{C}_{18}\text{H}_{18}\text{N}_4$, is generated by crystallographic twofold rotation symmetry, with two C atoms lying on the rotation axis. The pair of pyridyl rings form a dihedral angle of $78.4(2)^\circ$ and the dihedral angle between the pyridyl ring and the central benzene ring is $110.4(2)^\circ$. The molecules are linked into a one-dimensional chain by $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds.

Related literature

For a related structure, see: Zou *et al.* (2003). For background, see: Tao *et al.* (2000); Johnson (1976).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{18}\text{N}_4$
 $M_r = 290.36$
 Orthorhombic, *Pbcn*
 $a = 23.844(5)$ Å
 $b = 7.0493(14)$ Å
 $c = 9.2504(19)$ Å

$V = 1554.8(6)$ Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 295(2)$ K
 $0.38 \times 0.24 \times 0.20$ mm

Data collection

Rigaku R-Axis RAPID
 diffractometer
 Absorption correction: multi-scan
 (ABSCOR; Higashi, 1995)
 $T_{\min} = 0.972$, $T_{\max} = 0.979$

13887 measured reflections
 1764 independent reflections
 982 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.056$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.116$
 $S = 1.06$
 1764 reflections

101 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.12$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H1}\cdots\text{N1}^i$	0.86	2.24	3.0625 (19)	160

Symmetry code: (i) $x, -y + 1, z - \frac{1}{2}$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2587).

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supplementary materials

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1,3-Bis(2-pyridylaminomethyl)benzene

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Comment

Recently, N-heterocyclic complexes have attracted much attention in the area of coordination and supramolecular chemistry. Many one-dimensional, two-dimensional, and three-dimensional frameworks can be assembled through the metal coordination of pyridine-based bridging complexes (Tao *et al.*, 2000). Accordingly, we have designed and synthesized the title molecule, (I). The corresponding molecule (Zou *et al.*, 2003), with terephthalaldehyde instead of isophthalaldehyde, has a related structure.

The molecule of (I) (Fig. 1) is symmetrical with twofold rotation symmetry. The pair of pyridyl rings form a dihedral angle of 78.4°. And the dihedral angle between the pyridyl ring and the central benzene ring is 110.4°. The crystal structure is stabilized by the presence of intermolecular N—H...N hydrogen-bonding interactions (Table 1) resulting in a one-dimensional chain structure.

Experimental

To a solution of isophthalaldehyde in toluene was added a solution of 2-aminopyridine in toluene. The mixture was refluxed for 10 h, and a yellow precipitate was obtained. The solid product was reduced in absolute methanol by sodium borohydride. Colourless prisms of (I) were obtained by recrystallization from methanol with a yield of 80%.

Refinement

The H atoms were placed in calculated positions (C—H = 0.95–0.97 Å, N—H = 0.86 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$.

Figures

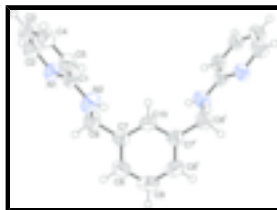


Fig. 1. A view of the molecular structure of (I). Displacement ellipsoids for the non-hydrogen atoms are drawn at the 50% probability level. Symmetry code: (i) $-x, y, 1/2 - z$.

1,3-Bis(2-pyridylaminomethyl)benzene

Crystal data

C₁₈H₁₈N₄

$F_{000} = 616$

supplementary materials

$M_r = 290.36$

Orthorhombic, *Pbcn*

Hall symbol: -P 2n 2ab

$a = 23.844$ (5) Å

$b = 7.0493$ (14) Å

$c = 9.2504$ (19) Å

$V = 1554.8$ (6) Å³

$Z = 4$

$D_x = 1.240$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 6970 reflections

$\theta = 3.0$ – 27.4°

$\mu = 0.08$ mm⁻¹

$T = 295$ (2) K

Prism, colourless

$0.38 \times 0.24 \times 0.20$ mm

Data collection

Rigaku R-AXIS RAPID
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 10.000 pixels mm⁻¹

$T = 295$ (2) K

ω scans

Absorption correction: multi-scan
(ABSCOR; Higashi, 1995)

$T_{\min} = 0.972$, $T_{\max} = 0.979$

13887 measured reflections

1764 independent reflections

982 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.056$

$\theta_{\text{max}} = 27.4^\circ$

$\theta_{\text{min}} = 3.0^\circ$

$h = -30 \rightarrow 30$

$k = -9 \rightarrow 9$

$l = -11 \rightarrow 10$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.116$

$S = 1.06$

1764 reflections

101 parameters

Primary atom site location: structure-invariant direct
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.046P)^2 + 0.1871P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.12$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.14189 (6)	0.6534 (2)	0.52951 (13)	0.0539 (4)
N2	0.11264 (6)	0.4662 (2)	0.33872 (15)	0.0634 (4)
H1	0.1127	0.4449	0.2472	0.076*
C1	0.14448 (6)	0.6112 (2)	0.38823 (16)	0.0475 (4)
C2	0.17287 (8)	0.8016 (3)	0.57373 (18)	0.0671 (5)
H2	0.1702	0.8362	0.6705	0.081*

C3	0.20785 (9)	0.9053 (3)	0.48820 (19)	0.0683 (5)
H3	0.2287	1.0056	0.5255	0.082*
C4	0.21115 (7)	0.8558 (3)	0.34387 (18)	0.0612 (5)
H4	0.2351	0.9212	0.2822	0.073*
C5	0.17922 (7)	0.7112 (3)	0.29261 (17)	0.0535 (5)
H5	0.1804	0.6788	0.1952	0.064*
C6	0.07853 (7)	0.3446 (3)	0.42870 (18)	0.0655 (5)
H6	0.0579	0.4210	0.4979	0.079*
H7	0.1024	0.2580	0.4821	0.079*
C7	0.03805 (7)	0.2334 (2)	0.33701 (17)	0.0508 (4)
C8	0.03766 (8)	0.0384 (3)	0.3357 (2)	0.0702 (5)
H8	0.0629	-0.0285	0.3929	0.084*
C9	0.0000	-0.0582 (4)	0.2500	0.0870 (10)
H9	0.0000	-0.1901	0.2500	0.104*
C10	0.0000	0.3274 (3)	0.2500	0.0521 (6)
H10	0.0000	0.4593	0.2500	0.063*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0647 (9)	0.0594 (9)	0.0375 (7)	-0.0036 (8)	-0.0005 (6)	-0.0021 (7)
N2	0.0714 (9)	0.0758 (11)	0.0429 (7)	-0.0207 (8)	0.0084 (7)	-0.0091 (8)
C1	0.0497 (9)	0.0526 (10)	0.0403 (8)	0.0039 (8)	-0.0027 (7)	-0.0012 (8)
C2	0.0911 (13)	0.0693 (13)	0.0410 (9)	-0.0153 (11)	-0.0001 (10)	-0.0067 (9)
C3	0.0903 (13)	0.0644 (12)	0.0501 (10)	-0.0194 (11)	-0.0062 (10)	-0.0014 (9)
C4	0.0708 (11)	0.0641 (12)	0.0486 (9)	-0.0100 (10)	-0.0030 (9)	0.0111 (9)
C5	0.0607 (10)	0.0633 (11)	0.0365 (8)	-0.0007 (9)	-0.0002 (8)	0.0010 (8)
C6	0.0687 (11)	0.0769 (14)	0.0509 (10)	-0.0132 (10)	-0.0037 (9)	0.0077 (9)
C7	0.0550 (9)	0.0503 (10)	0.0471 (9)	-0.0008 (8)	0.0030 (8)	0.0039 (8)
C8	0.0798 (12)	0.0558 (12)	0.0751 (13)	0.0114 (10)	-0.0052 (11)	0.0086 (10)
C9	0.116 (2)	0.0406 (16)	0.105 (2)	0.000	-0.014 (2)	0.000
C10	0.0662 (15)	0.0389 (13)	0.0512 (13)	0.000	0.0022 (12)	0.000

Geometric parameters (\AA , $^\circ$)

N1—C1	1.3416 (19)	C5—H5	0.9300
N1—C2	1.344 (2)	C6—C7	1.505 (2)
N2—C1	1.353 (2)	C6—H6	0.9700
N2—C6	1.445 (2)	C6—H7	0.9700
N2—H1	0.8600	C7—C8	1.375 (3)
C1—C5	1.402 (2)	C7—C10	1.3819 (19)
C2—C3	1.362 (2)	C8—C9	1.378 (2)
C2—H2	0.9300	C8—H8	0.9300
C3—C4	1.382 (2)	C9—H9	0.9300
C3—H3	0.9300	C10—C7 ⁱ	1.3819 (19)
C4—C5	1.358 (2)	C10—H10	0.9300
C4—H4	0.9300		
C1—N1—C2	116.33 (14)	N2—C6—C7	110.19 (14)

supplementary materials

C1—N2—C6	124.67 (14)	N2—C6—H6	109.6
C1—N2—H1	117.7	C7—C6—H6	109.6
C6—N2—H1	117.7	N2—C6—H7	109.6
N1—C1—N2	118.11 (14)	C7—C6—H7	109.6
N1—C1—C5	122.04 (15)	H6—C6—H7	108.1
N2—C1—C5	119.84 (14)	C8—C7—C10	118.02 (18)
N1—C2—C3	125.25 (16)	C8—C7—C6	122.00 (17)
N1—C2—H2	117.4	C10—C7—C6	119.98 (16)
C3—C2—H2	117.4	C7—C8—C9	120.2 (2)
C2—C3—C4	117.39 (18)	C7—C8—H8	119.9
C2—C3—H3	121.3	C9—C8—H8	119.9
C4—C3—H3	121.3	C8—C9—C8 ⁱ	120.7 (3)
C5—C4—C3	119.68 (17)	C8—C9—H9	119.6
C5—C4—H4	120.2	C8 ⁱ —C9—H9	119.6
C3—C4—H4	120.2	C7 ⁱ —C10—C7	122.7 (2)
C4—C5—C1	119.23 (15)	C7 ⁱ —C10—H10	118.6
C4—C5—H5	120.4	C7—C10—H10	118.6
C1—C5—H5	120.4		
C2—N1—C1—N2	-178.17 (15)	N2—C1—C5—C4	-179.68 (15)
C2—N1—C1—C5	2.4 (2)	C1—N2—C6—C7	165.56 (15)
C6—N2—C1—N1	-4.2 (2)	N2—C6—C7—C8	120.92 (19)
C6—N2—C1—C5	175.23 (16)	N2—C6—C7—C10	-58.67 (19)
C1—N1—C2—C3	-2.8 (3)	C10—C7—C8—C9	-0.3 (2)
N1—C2—C3—C4	1.0 (3)	C6—C7—C8—C9	-179.86 (13)
C2—C3—C4—C5	1.4 (3)	C7—C8—C9—C8 ⁱ	0.14 (12)
C3—C4—C5—C1	-1.7 (3)	C8—C7—C10—C7 ⁱ	0.13 (12)
N1—C1—C5—C4	-0.3 (2)	C6—C7—C10—C7 ⁱ	179.74 (16)

Symmetry codes: (i) $-x, y, -z+1/2$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H1 \cdots N1 ⁱⁱ	0.86	2.24	3.0625 (19)	160

Symmetry codes: (ii) $x, -y+1, z-1/2$.

Fig. 1

